

5 2 Review And Reinforcement Answers

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5 2 Review And Reinforcement

5-2 Review and Reinforcement - Steffin

4-5 Review and Reinforcement Electron Configurations If the statement is true, write "true" If it is false, change the underlined word or words to make the statement true ____ 1 The Pauli Exclusion Principle states that an orbital can hold a maximum of two electrons

5-2 Review and Reinforcement - Steffin

5 Cs-137 (t $\frac{1}{2}$ =302-years) is produced as a waste product in nuclear fission reactors What fraction (N/N o) remains undecayed after 2500-years? 10 If your cellar was measured to contain 24-g of Rn-222 (t $\frac{1}{2}$ =38-days) (a radioactive gas naturally produced by some granite deposits), how long would it take for that sample to decay to 02-g?

5 2 REVIEW AND REINFORCEMENT ANSWER KEY PDF

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5-2 Review and Reinforcement - MRS. Nairn

5-2 Review and Reinforcement Reading the Periodic Table On the line at the left, write the letter of the appropriate location of each group of elements on the periodic table below Some letters will be used more than once 1 carbon family 8 f-block elements 2 alkaline earth metals 9 noble gases 3 inner transition metals 10 p-block

2-5 Review and Reinforcement - MRS. Nairn

2-5 Review and Reinforcement Mixtures Use what you have learned in Section 2-5 to identify each of the following substances as pure substances,

heterogeneous mixtures, or homogeneous mixtures Write "pure," "heterogeneous," or "homogeneous" on the line 1 alphabet soup 2 salt 3 concrete 4 vegetable oil 5 air

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6—2 Review and Reinforcement Reading the Periodic Table On the line at the left, write the letter of the appropriate location of each group of elements on the Use the skills you developed in Section 5—2 to answer each of the following questions Below is the abbreviated electron configuration for sodium Explain each part of this

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5—3 Review and Reinforcement Periodic Trends 2 3 5 6 8 Li or K Ca or Ni Ga or B Cl or Br Be or Ba Si or S Fe or Au Which ion in each pair has the smaller atomic radius? 10 11 or 12 CO 13 14 Which atom or ion in each pair has the larger ionization energy? 15 16

2 Answer - Weebly

2 AnswerKey 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the dry cell battery The chemical energy in the 2-5 Review and Reinforcement, pages 28-29 1 heterogeneous 2 pure 3 heterogeneous 4 pure 5 homogeneous 6 heterogeneous 7 homogeneous 8 heterogeneous

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Name 7—2 Review and Reinforcement Covalent Bonding C ass If the statement is true, write "true" If it is false, change the underlined word r words to make it

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11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true" If it is false, change the underlined word or words to make it true Write your answer on the line provided 1 2 5 The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume

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24—1 Review and Reinforcement (continued) 13 There are 329 g of iodine-126 remaining in a sample originally containing 263 g of iodine-126 The half-life of iodine-126 is 13 days How old is the sample? 14 A sample that originally contained 25 g of rubidium-87 now contains 125 g The half-life of rubidium-87 is 6×10^{10} years

10-2 Review and Reinforcements - Front Page

10—2 Review and Reinforcement (continued) 9 00621 mol SS = (Q \$ gK2S 10 658gMg= 1 atoms Mg Solve each of the following problems as directed Show all your work 11 Calculate the number Of atoms in 040 mol of sulfur- 12 Calculate the number of atoms in 230 mol of ...

Name Date Class 3—3 Review and Reinforcement

3—3 Review and Reinforcement Modern Atomic Theory Complete the following table 1 2 3 5 In an individual, neutral atom, the number of protons always equals the number of electrons T 6 When an atom loses or gains one or more electrons, it is called an iZ 7 The average mass of an element's atoms is called the atomic number

REVIEW and REINFORCEMENT Section 6-3

REVIEW and REINFORCEMENT Chemical Families Section 6-3 KEY CONCEPTS - ~ Elements within the same family of the periodic table have simi(ar properties because they have the same number of valence electrons M Vocabulary Skills: Applying Definitions On the periodic table below,

label each of the following: acnoid series alkaline earth

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